

EVALUATION OF THE POWDER AND COMPACTION PROPERTIES OF MICROCRYSTALLINE STARCH (MCS) DERIVED FROM CASSAVA (MANIHOT ESCULENTA CRANTZ) STARCH BY ENZYMATIC HYDROLYSIS*Y.E. APEJI¹, A.R. OYI¹ AND H.MUSA¹*For author affiliations, see end of text*This paper is available online at www.jprhc.in**ABSTRACT**

The aim of this study was to investigate and determine the powder and compaction properties of microcrystalline starch (MCS) and compare with the properties of a well known direct compression filler-binder, microcrystalline cellulose (MCC).

Cassava starch was extracted from the freshly harvested tubers of *Manihot esculenta* Crantz and subjected to 5 hours of enzymatic hydrolysis to yield microcrystalline starch. The powder and compaction properties were evaluated and compared with MCC 101, a commercial brand of microcrystalline cellulose.

Results of the powder properties of MCS revealed differences in the particle size, angle of repose, flow rate, bulk density, tapped density, true density, Hausner's ratio, Carr's index and powder porosity when compared

to MCC. The compaction studies of both materials revealed that MCS had a faster onset of deformation and a greater extent of deformation in comparison to MCC. These results suggest that MCS has the potential of being used as a filler-binder in direct compression tableting.

Key words: Microcrystalline starch (MCS), Microcrystalline cellulose (MCC), Powder properties, Compaction properties, Direct compression tableting and Filler-binder.

INTRODUCTION

Tablet manufacturing by direct compression has advanced steadily over the years and has become the most preferred because of its simplicity, rapidity, being economical and of higher stability when compared to other dosage forms (5). This technology consists of compressing powdered blends of materials into tablets directly rather than going through the conventional granulation process which in turn avoids any change in the physical nature of the powdered material to be compressed (4).

The success of a direct compression process is influenced to a large extent by the functionality of the excipient chosen and this is determined greatly by its fundamental properties (powder and compaction properties). It therefore becomes necessary to characterize an excipient intended for use as a direct compression excipient before developing any formulation with that material by studying its powder and compaction properties in order to ascertain its functionality.

Starch is a versatile, cheap and readily available material obtained from renewable sources that has found wide application in tableting as a binder, disintegrant, diluent, lubricant and glidant. Unfortunately, it is not suitable for direct compression formulation due to its poor

compressibility and flow characteristics. There is a need therefore to impart these properties requisite for direct compression by modifying starch. Several authors have employed physical and chemical methods as well as coprocessing to develop excipients suitable for direct compression. The objective of this study is to modify starch by enzymatic hydrolysis using α -amylase enzyme in order to produce microcrystalline starch, a multifunctional excipient adaptable for direct compression tableting. The powder properties will be determined by evaluating the flow and compressibility profiles while compaction studies will be carried out using the models of Heckel and Kawakita equations. These properties will be studied in comparison to microcrystalline cellulose as a reference.

MATERIALS AND METHODS**MATERIALS**

The following materials used were all of pharmaceutical grade; Hydrochloric acid, A7595 α -amylase enzyme (Sigma-Aldrich laborchemikalien GmbH Germany), Xylene, Ethanol (95% v/v) (BDH Chemicals Ltd Poole, England), Sodium Hydroxide (Avondale laboratories Ltd Banbury, England) and Microcrystalline cellulose (MCC PH 101) (ATOZ

Pharmaceuticals Ltd Ambaltur, India). Cassava starch was extracted in the Process laboratory of the Department of Pharmaceutics & Pharmaceutical Microbiology, Ahmadu Bello University, Zaria.

METHODS

Synthesis of Microcrystalline starch

Cassava starch was extracted from freshly harvested tubers of *Manihot esculenta* Crantz using a method described in literature (3). The method described by Buwalda and Arends-Scholte (6) was adopted in the synthesis of microcrystalline starch.

One hundred grams (100g) of starch slurry was prepared containing 40%^{w/w} of cassava starch. It was then placed in a water bath (Digital thermostatic water bath) and the temperature set to 56°C. The pH of the reaction medium was adjusted to 6 using 0.1N and 0.2ml of α-amylase (BAN 240L) was dosed into the reaction mixture. The hydrolysis was allowed to run for 5hr with constant stirring.

After 5hr, the activity of the enzyme was terminated by lowering the pH to 3 with 0.1N HCl and subsequently neutralized by raising the pH to 7 with 0.1N NaOH. The reaction mixture was allowed to settle and the supernatant decanted. It was then washed several times with distilled water before adding 500ml of ethanol (95%^{v/v}) to dehydrate the microcrystalline starch (MCS) formed. The MCS was recovered by decanting the ethanol and then air-dried. It was then homogenised using a blender.

Investigation of Powder Properties

Moisture Content

Five grams (5g) of each powder sample was accurately weighed and dried at 105°C in the oven to constant weight. The sample was reweighed and weight loss calculated.

Angle of repose and Flow rate

The method described by Alebiowu (2) was adopted. Twenty grams (20g) of each powder sample was weighed and placed in a standing cylindrical tube. The cylinder was gently raised to leave a free heap of the powder. The circumference of the base of the heap was outlined and its radius, r, measured. The height of the heap, h, was also measured and the angle of repose calculated using the equation,

$$\theta = \text{Tan}^{-1} \frac{h}{r} \dots \dots \dots 1$$

The mean of three determinations was recorded. Flow rate of was determined by placing twenty grams (20g) of the powder in an Erweka flow apparatus (Type GDT, Erweka Apparatebau GmbH, West Germany) and allowed to flow through the funnel orifice. The time

taken for the powder to flow through the orifice was noted and the flow rate was determined as the ratio of weight (g) to time (seconds). The mean of three determinations were recorded.

True Density

The true density of the material was determined using a method described elsewhere (18). Xylene was employed as the displacement liquid using the pycnometer bottle.

Bulk and Tapped Densities

Exactly 50 gm of each powder sample was weighed and poured through a glass funnel into a 100 ml measuring cylinder at an angle of 45°. The cylinder was dropped on a wooden platform from a height of 2.5 cm three times at 2 second interval. The volume occupied by the powder recorded as the bulk volume. The cylinder was then tapped on the wooden platform until the volume occupied by the powder remained constant. This was repeated three times for both powders. The data generated were used in computing the Carr’s index and Hausner’s ratio for both powders.

Compressibility Index

The compressibility indices of Carr’s index and Hausner’s ratio were computed using the formulas below;

Carr’s index=.....2

Hausner’s ratio=.....3

Swelling, Hydration and Moisture sorption capacities

The swelling capacity of the powder was estimated by a method described by Iwuagwu and Onyekweli (12). The tapped volume occupied by 5 g of the powder, V_x, was noted. The powder was then dispersed in 85ml of water and the volume made up to 100 ml with more water. After 24 h of standing, the volume of the sediment, V_v, was estimated. The swelling capacity was determined using the formula, V_x/V_v. The method of Kornblum and Stoopak (15) was used to determine the hydration capacity. A 1g sample was placed in each of four 15ml plastic centrifuge tubes to which 10ml distilled water was added and then stoppered. The contents were mixed on a vortex mixer for 2min. The mixture was allowed to stand for 10min and then centrifuged at 1000rpm for 10min on a bench centrifuge. The supernatant was carefully poured out and the sediment weighed. The hydration capacity was determined as the ratio of sediment weight to the dry sample weight. Moisture sorption test was carried out using 2g of each material. It was weighed and evenly distributed over the surface of a 70mm tarred Petri-dish. The samples were placed in a dessicator containing distilled water in its reservoir (RH = 100%) at room temperature and the weight gained by the exposed samples at the end of the five day period was recorded and the amount sorbed was calculated from the weight difference.

Compaction Studies

Compacts of each material weighing 500mg were prepared by compressing them for 30s at various

Heckel equation classifies powders based on their compaction behaviour into three types; A, B and C (23). Type A materials are characterised by a straight line plot occurring over a wide range of pressures. The Heckel plot obtained for the study confirms that both materials are type A in nature.

The yield pressure, P_Y , resolved from the slope of the curve is a measure of the plasticity and is considered as that pressure at which yielding occurs during compaction (16). In general, a low P_Y value reflects low resistance to pressure, good densification and easy compression (11). The P_Y value obtained for MCS as presented on Table 2 was significantly lower compared to MCC indicating that MCS has a faster onset of deformation at low pressures. The D_0 values presented on Table 2 was determined by the ratio of the powder's loose bulk density to its particle density (19). This parameter represents the phase of densification occurring during compaction as a result of die-filling. It is the relative density of the powder at the point when the applied pressure is equal to zero. The greater the value of D_0 , the greater the degree of packing

or consolidation of the powder bed. MCS was shown to have a higher D_0 compared to MCC indicating that it has a greater degree of packing as confirmed by its bulk and tapped densities. The relative density, D_A , was calculated from the intercept using the equation,

$$D_A = 1 - e^{-A} \dots\dots\dots 6$$

While D_B is the relative density describing the phase of densification occurring during rearrangement of the powder bed at low pressure and is given as;

$$D_B = D_A - D_0 \dots\dots\dots 7$$

The values of D_A and D_B are presented on Table 2.

The Kawakita equation is a commonly used expression to linearize compression data. The equation describes the relationship between the degree of compression (C) of a bed of particles in a die and the applied pressure during compression (P) as follows;

$$\frac{P}{C} = \frac{P}{a} + \frac{1}{ab} \dots\dots\dots 8$$

Table 1: Powder Properties of MCS and MCC

Property	MCS	MCC
Angle of repose (°)	45.4 ± 1.03	46.2 ± 1.77
Flow rate (g/s)	1.5 ± 0.08	0.71 ± 0.02
Bulk density (g/ml)	0.61 ± 0.02	0.39 ± 0.02
Tapped density (g/ml)	0.79 ± 0.02	0.55 ± 0.01
True density (g/ml)	1.38	1.48
Hausner's ratio	1.30	1.41
Carr's Index (%)	23	29
Swelling capacity	1.50	1.31
Hydration capacity	0.82	0.84
Moisture sorption capacity (%)	19	10

Table 2: Compact Parameters obtained from Heckel and Kawakita plots

Material	Heckel Parameters				Kawakita Parameters			
	P_Y	D_0	D_A	D_B	a	b	D_I	P_K
MCS	45.45	0.413	0.933	0.52	0.562	0.809	0.438	1.24
MCC	111.11	0.264	0.874	0.61	0.735	0.673	0.265	1.49

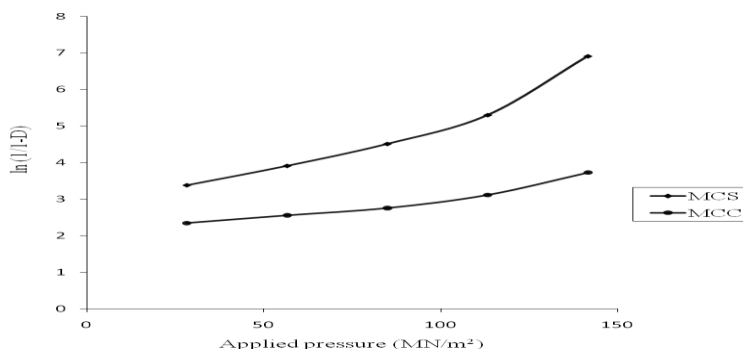


Figure 1: ln(1/(1-D)) against applied pressure for MCS and MCC compacts

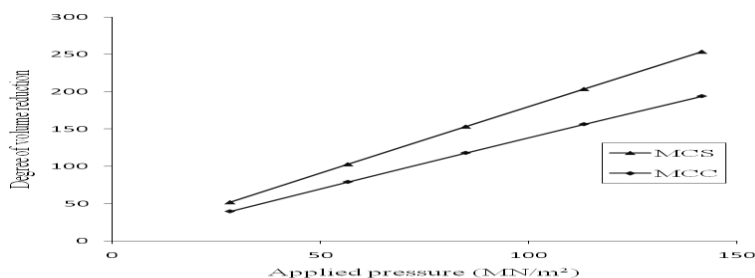


Figure 2: Degree of volume reduction against applied pressure (Kawakita plots for MCS and MCC)

The Kawakita plot is displayed as Figure 3 in the study. A straight line curve was obtained at all pressures linearizing the compression data. The compressibility parameters, a & b were obtained from the slope and intercept of the plot. The compression parameter, a, reflects the total degree of volume reduction of the powder at infinite applied pressure. The value recorded for MCS on Table 2 was greater when compared to MCC. This suggests that MCS achieves a closely packed structure with minimal porosity necessary for effective bonding compared to MCC. The reciprocal of the compression parameter, b (i.e. 1/b) denoted as P_K represents an indication of the stress/pressure at which particles deform or fail during confined compression (1, 13 and 17). The P_K values observed for both powders (Table 2) shows that MCS deforms at a lower pressure

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compared to MCC. Plastic deformation of the particles in the bed is brought in close proximity to each other facilitating bond formation resulting in tablets of sufficient mechanical strength. This is a common feature of many filler-binders in direct compression that they undergo plastic deformation during compaction (16).

CONCLUSION

The powder and compaction properties of MCS compared well with MCC as revealed by the study and so can be incorporated as a directly compressible excipient in tablet formulations but precautions should be taken with moisture sensitive drug formulations by including handy packs of dessicants.

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