# Metal distribution in the sediments, water and naturally occurring macrophytes in the river Gomti, Lucknow, Uttar Pradesh, India

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River Gomti receives treated/untreated industrial as well as municipal wastes from various drains of Lucknow city, India. In order to study heavy metal pollution (Cd, As, Pb and Cu) in the river, water and sediment samples were collected from 10 sampling stations along a 9 km stretch in the city of Lucknow. Results revealed that the concentrations of heavy metals in water samples were in the range: As: 0.035-0.061, Cd: 0.016-0.068, Cu: 0.029-0.062 and Pb:  $0.031-0.065 \text{ mg l}^{-1}$  whereas in sediments metal concentrations were found to be As: 3.72-14.98, Cd: 1.91–8.39, Cu: 8.97–95.35 and 35.82–90.92  $\mu$ g g<sup>-1</sup>. Bioaccumulation of these metals was assessed in four aquatic macrophytes, viz. Pistia stratiotes, Eichhornia crassipes, Polygonum coccineum and Marsilea quadrifolia. Pistia stratiotes and Polygonum coccineum accumulated maximum amount of Pb followed by Cu, Cd and As, whereas in the case of Eichhornia crassipes and Marsilea quadrifolia the relative metal accumulation pattern was found as Cu > Cd > Pb > As and Cu > Pb > Cd > As respectively. The present study suggests that though the concentrations of toxic metals were lower in water, chronic exposure could result in bioaccumulation to a degree many-fold higher than in growing medium. It was also concluded that the water and sediment of the river should be regularly monitored for heavy metal contamination and care should be taken while using river water in agriculture/ aquaculture.

Keywords: Bioaccumulation, Gomti River, heavy metals, macrophytes, sediments.

MOST civilizations in the world flourished on or close to river banks. As a direct consequence and through passage of time, the water and the sediments of these rivers suffered contamination affected by municipal and industrial contaminants<sup>1</sup>. Studies have shown that toxic metals constituted significant proportion in these waste materials which impart acute and chronic virulent characteristics to the wastes and sediments<sup>2</sup>. Heavy metals present in water, sediments and macrophytes play a crucial role in identifying the sources and paths of heavy metal pollution in aquatic ecosystems<sup>3–5</sup>. River water quality broadly depends on natural processes such as precipitation, soil erosion, weathering, etc. and human input, mainly through exploitation of water for industrial, agricultural and urban activities<sup>6,7</sup>. Because river water is used as a primary water resource for irrigation and aquaculture, it is imperative to have a reliable knowledge about the quality of water for its wise and judicious utilization.

Earlier river sediments were considered as a depot of various chemical species; they are now recognized as active aquatic compartment which performs the principal role in redistribution of deposited chemical species to water and aquatic biota<sup>8,9</sup>. Studies have demonstrated that heavy metal concentrations in sediments are several orders of magnitude higher than their concentration in water and act as secondary source of metals to the aquatic environment<sup>10</sup>. Hence, metal deposition in river water and sediments should be regularly monitored to assess the overall health and suitability of water for various purposes<sup>11–15</sup>.

Aquatic macrophytes growing in the natural ecosystem can be used to assess the overall health and past trends of aquatic environment<sup>2</sup>. Furthermore, being immobile in nature these macrophytes have developed the ability to accumulate toxic metals in their body parts, which makes them an effective bio-indicator of metal pollution<sup>16</sup>. Studies demonstrate that uptake of metal depends on the speciation of metal and the type of macrophytes; submerged, rooted submerged, floating and rooted merged. Further, the distribution, mobility and availability of metals to these plants are influenced by physicochemical characteristics of water<sup>16,17</sup>.

In view of continuous interaction among the sediment, water and aquatic biota, a simultaneous evaluation of metal concentration in these segments is essential for

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Figure 1. Sampling sites on river Gomti, Lucknow, India.

generating reliable information on the fate and behaviour of metal in a river ecosystem. As the river fulfils most needs of irrigation water, the contamination level needs to be monitored regularly. Further, to design an efficient pollution control and water resource management programme, data of this type is important.

The Gomti river, one of the major tributaries of Ganga, originates from a reservoir in the swampy and densely forested area near Madho-Tanda (Miankot, elevation of about 200 m amsl, 28°34'N and 80°07'E) about 3 km east of Pilibhit in Uttar Pradesh and about 50 km south of the Himalayan foot hills. The river flows in the great Gangetic alluvial plain, which is of Pleistocene-Holocene age, and redistributes the weathered sediments of the Gangetic alluvial plain derived from Himalayas. The river flowing through the districts of Pilibhit, Sahjahanpur, Jaunpur and Ghazipur in Uttar Pradesh covers a distance of about 730 km before finally joining the river Ganga near Udyarghat in Ghazipur district, about 30 km north of Varanasi. The domestic water demand of Lucknow city is partially met by the Gomti river water, delivered from Gaughat upstream of Lucknow. The river subsequently receives approximately 450 MLD of untreated domestic waste water from Lucknow city in addition to various point and non-point discharges from agricultural run-off. Like many other important rivers of the country Gomti is characterized by a sluggish flow except in the monsoon period. Sediments of the Gomti river are characterized by fine sand with slight changes along the course of the river<sup>18</sup>.

Several studies have assessed the metal contamination in water<sup>3,18–22</sup> and sediments<sup>3,20</sup> of river Gomti (Table 1). However, as far as we are aware, no comprehensive study has been conducted to monitor the metal contamination simultaneously in river water, sediments and naturally grown macrophytes. In the present study, we have assessed heavy metal concentration in water, sediments and macrophytes simultaneously and toxic metals in the river ecosystem.

#### Materials and methods

#### Site description

Sampling was carried out at 10 sites along river Gomti in Lucknow city (Figure 1). The chosen sites were: Gau Ghat, Kuriya Ghat, Daliganj, Shaheed Smarak, Hanuman Setu, Boat Club, Lakshman Mela Ground, Khatu Shyam Vatika, Baikunth Dham and Gomti Baragge (Figure 1). A sampling network was designed to account for inputs from waste water drains that have an impact on river quality.

#### Sample collection and analysis

#### Water

Water samples were collected from the river bank in polyethylene bottles. The physicochemical parameters were

determined following standard methods for analysis of water and waste water<sup>23</sup>. For physicochemical parameters, samples were preserved by adding an appropriate reagent and brought to the laboratory in sampling kits maintained at 4°C for analysis. The unfiltered water samples for total metal analysis were preserved using ultrapure nitric acid to lower the pH to <2.0. Samples were digested on a hot plate using acid mixture (HNO<sub>3</sub>: HClO<sub>4</sub>: 5:1). The digested samples were then filtered and the final volume was made up to 25 ml with deionized water and analysed on atomic absorption spectrophotometer (AA 240 FS, Varian)<sup>2,22</sup>.

#### Sediment

The bed sediments of the Gomti river were collected at 10 sampling sites. The sediment samples were transferred to wide mouth 500 cm<sup>3</sup> polypropylene bottles. The samples were air-dried and ground in a porcelain pestle mortar. Samples of the bed sediments  $(1.0 \pm 0.05 \text{ g})$  were digested with concentrated HNO<sub>3</sub>: HClO<sub>4</sub> (5:1). The final volume was made up to 25 ml and analysed by atomic absorption spectrophotometer.

#### Macrophytes

Four aquatic macrophytes, viz. Pistia stratiotes, Eichhornia crassipes, Polygonum coccineum and Marsilea quadrifolia, were collected from the same sites along the river bank from where water and sediment samples had been collected. These aquatic plants were brought to the laboratory and then washed with running tap water followed by distilled water to remove extraneous matter. After washing, the plant material was oven-dried at 65°C till constant weight. For metal estimation, one gram plant tissue was taken and digested in a solution of HNO<sub>3</sub>:  $HClO_4$  (3:1) at 70–80°C. The solution was allowed to evaporate by increasing the temperature to 105°C until the solution became transparent. The final volume was diluted to 25 ml with 0.1 N HNO<sub>3</sub>, filtered through 0.25 µm filter paper and analysed on atomic absorption spectrophotometer<sup>24,25</sup>. All readings were taken in six replicates (n = 6) and the results are expressed as mean  $\pm$  SE.

#### Enrichment coefficient and translocation factor

Enrichment coefficient (EC) was determined to derive the degree of heavy metal accumulation in plants growing on contaminated environment<sup>26,27</sup>

$$EC = \frac{Concentration of metal in roots/shoots}{Concentration of metal at contaminated site}$$

Translocation factor (TF) or mobilization ratio of each metal was calculated to determine the transloca-

$$TF = \frac{Concentration of metal in plant shoots}{Concentration of metal in plant roots}$$
.

#### **Results and discussion**

#### Water

Each water body possesses a specific pattern of physicochemical characteristics which is primarily governed by the climate, geomorphology and geochemistry prevalent in that river basin. The results of physicochemical analysis and heavy metal concentrations of water are presented in Tables 2 and 3. In the present study, the pH of water ranged between 6.54 and 8.14. The pH of river water was found to be alkaline at all sites except at two locations viz. Daliganj and Hanuman Setu;  $6.54 \pm 0.29$  and  $6.71 \pm 0.92$  respectively. Alkaline pH is known to favour the bioavailability of metals<sup>28</sup>. The mean value of dissolved oxygen (DO) varied between 3.69 and 7.3 mg  $l^{-1}$ . The maximum DO was found in the upstream section, i.e. Gaughat. In the downstream path it was depleted. This may be due to discharge of municipal and industrial effluents into the river<sup>29</sup>. The chloride and sulphate content in water ranged between 4.07–8.8 and 16.04–26.03 mg  $l^{-1}$ respectively, and found to be within the tolerance limit given by the Bureau of Indian Standards (BIS), 2012 (ref. 30) (chloride = 600 and sulphate =  $1000 \text{ mg l}^{-1}$ ). These ions may have become part of river water due to interaction with the soil system and/or due to use of various inorganic chemicals discharged from industries and agriculture land<sup>2,31</sup>. Heterogeneous hardness of water was recorded at different sampling sites. Ca<sup>++</sup> and Mg<sup>++</sup> are the principal cations imparting hardness to water; however, Fe<sup>++</sup> and Mn<sup>++</sup> may also contribute cations. The anions influencing the hardness are mainly carbonate, bicarbonate, sulphate, fluoride and silicate<sup>32</sup>. Solid particles are natural constituents of river water systems and are introduced by both natural and anthropogenic sources. At all sampling locations, the value of total dissolved solids was found below the tolerance limit (2100 mg  $l^{-1}$ ) given by BIS (ref. 30). The maximum nitrate concentration was found at site 3. The high concentration of nitrate along the stream in the city may be attributed to the discharge of agricultural and domestic waste water into the river<sup>18</sup>. The value of nitrite was found below 1 mg  $l^{-1}$  at sites other than 3 and 4, which may be due to discharge of industrial effluents near these sampling sites. High nitrite concentration indicates contamination due to industrial effluents with unsatisfactory microbial quality of water<sup>11</sup>. The values of heavy metals found in water at different sites are presented in Table 3.

								Sampling s	ites					
		Gau Ghat		Mohan Dalij	Maikin/ ganj	Martyr me	morial	Hanu	man Setu	Nishat Khatu Shya	tganj/ am Vatika	Pipr	a Ghat	
fetal Ye	ar W	ater Sedi	iment	Water	Sediment	Water S	sediment	Water	Sediment	Water \$	Sediment	Water	Sediment	Reference
20 20 20 20 20 20 20 20 20 20 20 20 20 2	05 0. 06 0. 07 0.	012 14 015 015 017 016	1.76	0.014 0.021 0.025 0.027	50.23	0.014 0.019 0.024 0.026	79.54	0.015 0.022 0.028 0.030	48.18	0.016 0.023 0.030 0.032	33.28	0.014 0.029 0.033 0.036	4.36	3, 18–20
20	14 0.	029 8.	.97	0.053	38.72	0.042	41.06	0.036	25.89	0.038	47.59		I	Present work
20 20 200 200	05 0 06 B 07 B 08 B 037 B	0001 1. .DL DL DL	.64	0.0002 BDL BDL BDL	17.26	BDL BDL BDL	16.01	BDL BDL BDL BDL	11.98	BDL BDL BDL	4.61	0.0003 BDL BDL BDL	2.25	3, 18–20
20	14 0.	016 1.	.91	0.023	7.14	0.017	7.12	0.024	5.87	0.021	7.37	I	I	Present work
b 200 200 200	05 0. 06 0 07 0 037 0 037 0	032 14 .15 .19 .26	1.21	0.030 0.26 0.31 0.36	22.28	0.024 0.042 0.047 0.053	51.11	0.005 0.037 0.041 0.048	38.94	0.007 0.011 0.015 0.021	40.04	0.027 0.015 0.019 0.026	5.80	3, 18–20
20	0 0	031 35	5.82	0.055	70.1 Table 2. Pl	0.047 hvsicochemica	73.43 I quality of	0.038 f Gomti rive	60.46 r water. Lucknow.	0.043 India	86.73	1	1	Present work
sampling site	s	Ηd	DC	0	Chloride	Hardness	4	TS	TSS	TDS		Sulphate	Nitrate	Nitrite
jau Ghat kuriya Ghat Daliganj shaheed Sma Hanuman Set Joat Club akshman Mc katu Shyam Saikunth Dha	rak u sla Ground un un	$7.62 \pm 0.52$ $8.05 \pm 0.13$ $8.05 \pm 0.13$ $6.54 \pm 0.29$ $7.74 \pm 0.43$ $6.71 \pm 0.243$ $7.18 \pm 0.243$ $7.515 \pm 0.02$ $7.55 \pm 0.12$ $7.56 \pm 0.05$ $8.14 \pm 0.01$	7.3 ± 6.4 ± 6.4 ± 7.3 ± 6.4 ± 7.9 ± 4.55 ± 4.95 ± 4.15 ± 4.15 ± 4.15 ± 3.96 ± 3.59 ±	0.15 0.22 0.05 0.13 0.05 0.01 0.05 0.15 0.15 0.05	$\begin{array}{c} 8.8 \pm 1.31\\ 2.8.4 \pm 2.10\\ 46.07 \pm 7.1\\ 21.3 \pm 2.03\\ 21.3 \pm 2.03\\ 37.285 \pm 3.8\\ 21.3 \pm 2.01\\ 14.66 \pm 1.72\\ 19.32 \pm 2.35\\ 19.32 \pm 2.35\\ 28.62 \pm 1.02\\ \end{array}$	$\begin{array}{c} 187.09 \pm 3.2 \\ 168 \pm 5.1. \\ 168 \pm 5.1. \\ 217.01 \pm 10. \\ 160 \pm 12. \\ 180.7 \pm 15. \\ 201 \pm 11. \\ 224 \pm 12. \\ 207 \pm 21. \\ 183.2 \pm 14. \\ 213.06 \pm 16. \end{array}$	8 61   54 680.   .1 693.   16 867   16 867   63 859.   63 859.   63 859.   63 853.   63 853.   63 853.   63 853.   63 853.   63 853.   63 853.   63 853.   63 853.   63 853.	$\begin{array}{c} 9 \pm 45.1 \\ 5 \pm 32.1 \\ 5 \pm 32.1 \\ 5 \pm 60.2 \\ 7 \pm 41.2 \\ 9 \pm 31.4 \\ 5 \pm 47.1 \\ 5 \pm 78.1 \\ 5 \pm 78.1 \\ 5 \pm 66.1 \\ 33 \pm 67.3 \\ 82 \pm 71.7 \end{array}$	$264 \pm 10.14$ $339.5 \pm 18.71$ $441.6 \pm 32.06$ $388 \pm 26.21$ $378.83 \pm 21.1$ $378.83 \pm 21.1$ $398.5 \pm 26.2$ $487 \pm 21.73$ $465 \pm 21.13$ $467.2 \pm 32.17$ $596.2 \pm 48.77$	355 ± 20. 355 ± 20. 409 ± 12. 468.7 ± 20 579 ± 37 508.01 ± 35 461 ± 41 484.5 ± 3 423.5 ± 3 478.45 ± 4 601.6 ± 6,	116 2: 22 116 2: 22 1.16 22 1.16 22 1.125 2: 1 1.125 2: 22 1.1.25 2: 23 1.1.25 2: 22 1.1.25	$\begin{array}{c} 4.77\pm2.50\\ 4.77\pm2.50\\ 1.643\pm1.05\\ 5.034\pm1.40\\ 9.82\pm2.05\\ 4.57\pm4.20\\ 5.37\pm2.32\\ 6.04\pm1.91\\ 1.05\pm1.21\\ 1.05\pm1.21\\ 3.712\pm2.20\end{array}$	$\begin{array}{c} 41.106\pm5.8\\ 39.63\pm3.52\\ 48.04\pm1.26\\ 51.049\pm3.8\\ 60.87\pm5.02\\ 57.934\pm5.6\\ 79.57\pm3.09\\ 44.288\pm4.7\\ 53.72\pm5.08\\ 61.04\pm2.05\\ 61.04\pm2.05\end{array}$	$\begin{array}{c} 0.474\pm0.5\\ 0.418\pm0.03\\ 0.307\pm0.04\\ 0.307\pm0.04\\ 0.201\pm0.17\\ 0.165\pm0.03\\ 0.159\pm0.07\\ 0.645\pm0.07\\ 0.214\pm0.02\\ 0.42\pm0.03\\ 0.38\pm0.03\end{array}$

**Table 3.** Heavy metals concentration (mg  $l^{-1}$ ) in the water of Gomti river, Lucknow, India

Sampling sites	Cd	As	Cu	Pb
Gau Ghat	$0.016\pm0.002$	$0.035 \pm 0.010$	$0.029 \pm 0.004$	$0.031 \pm 0.009$
Kuriya Ghat	$0.019\pm0.002$	$0.040 \pm 0.004$	$0.033 \pm 0.002$	$0.035 \pm 0.003$
Daliganj	$0.023 \pm 0.004$	$0.049 \pm 0.011$	$0.053 \pm 0.014$	$0.055 \pm 0.004$
Shaheed Smarak	$0.017 \pm 0.005$	$0.037 \pm 0.006$	$0.042 \pm 0.007$	$0.047 \pm 0.007$
Hanuman Setu	$0.024\pm0.004$	$0.041 \pm 0.009$	$0.036\pm0.005$	$0.038\pm0.007$
Boat Club	$0.043 \pm 0.019$	$0.043 \pm 0.007$	$0.049\pm0.005$	$0.037\pm0.006$
Laxman Mela Ground	$0.061 \pm 0.019$	$0.057 \pm 0.009$	$0.039\pm0.006$	$0.044 \pm 0.001$
Khatu Shyam Vatika	$0.021 \pm 0.003$	$0.043 \pm 0.004$	$0.038 \pm 0.007$	$0.043 \pm 0.002$
Baikunth Dham	$0.039\pm0.006$	$0.044 \pm 0.006$	$0.050 \pm 0.009$	$0.045 \pm 0.005$
Gomti Baragge	$0.068\pm0.002$	$0.061\pm0.009$	$0.062\pm0.004$	$0.065\pm0.005$

The data are represented by means of six replicates  $\pm$  SE.

**Table 4.** Concentration of heavy metals  $(\mu g g^{-1})$  in sediments of Gomti river

Sampling sites	Cd	As	Cu	Pb
Gau Ghat	$1.907 \pm 0.17$	$3.723\pm0.86$	$8.967 \pm 1.02$	$35.823 \pm 4.73$
Kuriya Ghat	$2.740 \pm 0.96$	$10.913 \pm 1.38$	$10.283\pm1.48$	$40.187 \pm 5.71$
Daliganj	$7.137 \pm 0.89$	$8.873 \pm 0.57$	$38.723 \pm 3.79$	$70.097\pm8.29$
Shaheed Smarak	$7.117 \pm 0.88$	$9.047 \pm 1.08$	$41.063 \pm 2.16$	$73.434 \pm 7.79$
Hanuman Setu	$5.863 \pm 0.09$	$6.207 \pm 0.54$	$25.887\pm3.80$	$60.463 \pm 4.01$
Boat Club	$4.484 \pm 0.53$	$11.113 \pm 1.97$	$48.817 \pm 3.51$	$57.910 \pm 9.72$
Laxman Mela Ground	$2.710\pm0.26$	$6.900\pm0.50$	$53.843 \pm 3.84$	$81.630\pm9.64$
Khatu Shyam Vatika	$7.370\pm0.73$	$8.024 \pm 0.38$	$47.590 \pm 4.37$	$86.727 \pm 8.31$
Baikunth Dham	$5.913 \pm 0.79$	$8.557 \pm 1.15$	$61.650 \pm 9.12$	$72.453 \pm 8.91$
Gomti Baragge	$8.390 \pm 0.83$	$14.980\pm3.02$	$95.350\pm7.83$	$90.920\pm7.82$

The data are represented by means of six replicates  $\pm$  SE.

The maximum concentration of all metals (Cu, Cd, As and Pb) was found at site 10, i.e. Gomti Barrage. The water was almost stagnant at this location, which might be responsible for high concentration of these metals. The Cd concentration in the river water ranged between 0.016 and 0.068 mg l<sup>-1</sup>. Cadmium is a toxic element and is introduced into the water body by metal industries, plastic industries, domestic waste, fossil fuel and sewerage system<sup>33</sup>. Arsenic concentration ranged between 0.035 and  $0.061 \text{ mg l}^{-1}$ . Although the prime sources of arsenic in the environment are minerals and various geo-genic processes, human activities such as use of arsenic containing pesticides, burning of fossil fuel, mining activities, use of fertilizers and wood preservatives also add to arsenic abundance in the environment<sup>34,35</sup>. Though Cu is a micronutrient at higher concentrations, it is toxic to plant and animals. The concentration of Cu falls in the range 0.029 to 0.062 mg l<sup>-1</sup>. Cu in river water might have been introduced along with waste discharged from automobiles, repair shops, electroplating units, utensil manufacturing units and workshops. Lead is a proven cumulative poison<sup>36</sup>. The concentration of Pb is in the range between 0.031 and 0.065 mg l<sup>-1</sup>. The presence of Pb in water samples from all the sampling sites indicates the deposition of Pb particles from vehicular exhaust. Agricultural run-off may also have contributed to the observed levels of Pb because it can occur as impurity in fertilizers, pesticides, compost and manure<sup>10,18</sup>. The present findings are in good agreement with earlier studies on river Gomti<sup>11,19,20</sup>. When the present results were compared with those from earlier studies, it was observed that there was cumulative increase in the concentration of Cu, Cd and Pb in the water of Gomti river (Table 1). Arsenic was not reported earlier in the water of river Gomti.

#### Sediment

The results of heavy metal analysis in the sediments of river Gomti are given in Table 4. All the metals studied were found to be present in high concentration in the river bed sediments at all sites. This may be due to sluggish flow of the river which allows particles to settle down. Total metal concentration in the sediments of river Gomti was found to be: Cd: 1.907-8.390, As: 3.723-14.980, Cu: 8.967–95.390 and Pb:  $35.823-90.920 \ \mu g \ g^{-1}$  (Table 4). The concentration of metals found in sediments are critically higher than that found in the river water. The alkalinity of river water is considered to favour the settling down of metals in the sediment. Higher concentration of metals in the river bed sediment subsequently poses high risk to benthic biota of the river due to metal toxicity. The occurrence of metals in sediments may be due to discharge of municipal and industrial effluents from various sources including untreated sewage, vehicular exhaust and agro-chemical run-off. The average concentration of

Sampling sites		Pistia	stratiotes			Е. сп	assipes		$P_G$	) mnuogh	soccineum		Μ	. quadrif	olia	
$\mathrm{EC}_{\mathrm{R}}$	Cd	As	Pb	Cu	Cd	As	Pb	Cu	Cd	As	Ъb	Cu	Cd	As	Ъb	Cu
Gau Ghat	477.58	30.39	507.39	278.27	538.32	87.96	250.01	376.65	74.23	16.2	320.35	215.34	33.11	8.32	45.98	129.42
Kuriya Ghat	321.58	84.98	845.96	346.93	561.58	143.95	242.11	321.94	64.95	24.28	215.91	265.87	48.00	17.22	29.83	173.57
Daliganj	533.21	82.83	408.80	301.32	753.97	206.12	164.33	266.54	214.90	67.13	304.39	318.57	45.00	22.44	52.79	114.44
Shaheed Smarak	465.09	365.75	725.53	648.65	846.98	239.22	279.22	143.12	178.31	90.13	298.51	289.99	232.03	38.12	43.51	115.75
Hanuman Setu	847.40	91.51	549.65	530.79	660.29	164.68	471.13	475.90	84.97	60.17	487.98	280.33	43.40	50.34	94.49	168.22
Boat Club	152.79	47.31	80.10	204.49	280.88	278.08	43.40	443.33	31.09	21.51	33.14	172.24	46.79	60.78	10.90	41.51
Lakshman Mela Ground	279.34	53.65	497.94	333.07	161.03	159.53	279.42	311.97	68.22	53.89	220.24	231.44	34.08	15.74	47.52	23.39
Khatu Shyam Vatika	291.29	48.75	629.08	315.13	439.50	189.38	200.70	600.34	177.74	56.32	345.77	196.43	61.55	47.18	19.94	131.53
Baikunth Dham Gomti Baragge	336.47 345.96	134.27 99.07	354.26 263.35	238.40 419.84	411.98 311.23	180.07 213.82	265.70 249.43	536.84 393.26	130.19 100.36	92.42 68.59	367.35 279.40	205.66 339.81	51.98 45.24	46.79 18.00	110.76 62.78	64.66 64.84
}																
ECs Gau Ghat	355 40	170 66	530 57	17276	331.00	58 786	157 18	70802	76 76	757	730 35	138 73	07 60	4 80	79.67	03 45
Vau Ullat Kuriva Ghat	04.000 038.47	116.22	703.85	376.63	00.166 11 663	00.200 101 32	01.201	162 76	40.70 51.05	15.66 15.66	00.002 186.43	185.08	43 37	4.00 0 13	20.40 20.40	0.4.07
Daligani	479 12	17.00	432.11	26536	795 00	264 25	199 16	2.08.58	135 00	26.05	254 02	34136	7C.CF	1732	23 31	95.05
Shaheed Smarak	448 33	2.67.61	389.08	560.88	791.69	187 19	231 28	70.64	118 69	55 12	20.26	241.63	185.08	2633	28.85	72,02
Hanuman Setu	865.81	34.15	444.00	380.09	535.07	111.32	340.43	504.15	76.93	49.61	407.58	251.83	74.79	30.39	52.90	134.51
Boat Club	129.07	70.15	51.17	188.37	253.95	264.92	29.811	388.37	11.63	14.65	38.32	45.85	42.37	24.12	3.49	22.29
Lakshman Mela Ground	219.02	70.06	697.41	334.21	108.70	140.65	249.47	405.97	47.12	52.29	250.61	217.52	16.65	6.92	38.48	21.05
Khatu Shyam Vatika	127.74	45.23	662.38	307.04	368.71	185.86	208.22	518.40	89.52	25.01	336.77	65.48	43.65	39.45	14.40	94.72
Baikunth Dham	333.62	105.14	263.38	280.60	415.94	196.74	243.15	361.74	126.54	61.58	265.95	179.64	26.64	25.72	52.63	57.42
Gomti Baragge	324.59	79.02	202.27	353.71	267.00	160.98	283.30	355.85	83.11	48.85	217.55	291.00	28.52	13.44	56.83	32.77
EC <sub>R</sub> , Enrichment coeffici	ent root, ECs	s, Enrichme	ent coefficien Table	it shoot.	slocation fac.	tor of differe	nt macronhv	tes collecte	d from the	Gomti Riv	'er					
		Pistia stra	tiotes		7	J. crassipes			Polveon	um coccin	eum		M.	auadrifo	lia	
						•			)		i					
Sampling sites	Cd	As	Pb (	Cu	Cd A	.s Pt	Cu	Cd	As	ld	C	СČ	1 F	As	Pb	Cu
Gau Ghat	0.744	0.758	1.044 0.	621 0.	.615 0.6	.0 <u>6</u>	9 0.792	0.657	0.467	0.7	19 0.6	44 0.80	53 0.5	577 0.	.645	.722
Kuriya Ghat	0.741	0.650	0.968 0.	941 0.	.930 0.7	0.75	<b>91 0.506</b>	0.786	0.645	0.80	53 0.6	96 0.9(	0.5	530 0.	.684	.816
Daliganj	0.899	0.958	1.057 0.	881 1.	.054 1.1	82 1.21	12 0.783	0.628	0.388	3 0.8	40 1.0	72 0.2	16 0.7	772 0.	442	.831
Shaheed Smarak	0.965	0.772	0.888 0.	865 0.	.935 0.7	82 0.82	28 0.494	0.666	0.612	0.0	58 0.8	33 0.7:	52 0.6	591 0.	.663	.622
Hanuman Setu	1.022	0.417	0.808 0.	716 1.	.002 0.6	576 0.72	23 1.059	0.905	0.824	0.8	35 0.8	98 1.7.	23 0.8	894 0.	560	.795
Boat Club	0.845	0.722	0.788 0.	921 0.	.904 1.6	136 0.68	39 0.876	0.374	0.681	1.1:	56 0.2	96 0.91	0.6	525 0.	.810	.537
Lakshman Mela Ground	0.784	1.306	1.401 1.	005 0	.675 0.8	82 0.89	33 1.301	0.691	0.676	1.1.	38 0.9	40 0.48	89 0.4	139 0.	.810	.900
Khatu Shyam Vatika	0.439	0.928	0.953 0.	974 0	.839 0.5	<b>8</b> 1 1.02	37 0.786	0.504	0.444	6.0 1	74 0.3.	33 0.7(	.0 0.	538 0.	722	.720
Baikunth Dham	0.992	0.783	0.743 1.	177 1	.010 1.6	1 <u>9</u> 3 0.91	15 0.822	0.972	0.652	0.7.	24 0.8	73 0.5	12 0.5	550 0.	475	.888
Gomti Baragge	0.938	0.798	0.768 0.	842 0	.858 0.7	53 1.15	36 0.815	0.828	0.712	0.8	52 0.8	56 0.6.	31 0.7	747 0.	.905	.505

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different metals in the Gomti river sediments during 1994–1998 was reported as: Cd: 0.092–17.83, Pb: 1.56–4.86, Cu: 3.60–245.33  $\mu$ g g<sup>-1</sup> (ref. 11). In another study, the heavy metal concentration in Gomti river sediments was reported as Cd: 1.64–17.26, Pb: 5.8–51.11, Cu 4.98–47.14 in summer; Cd 1.20–5.44, Pb 10.77–40.33, Cu: 4.36–79.54, in winter and Cd 0.41–12.96, Pb 29.63–100.38, Cu 3.98–91.17  $\mu$ g g<sup>-1</sup> in rainy seasons<sup>19</sup>. The concentration of metals in the sediment obtained in the present study is comparable to that reported earlier<sup>11,20</sup>.

#### Macrophytes

In freshwater ecosystem, macrophytes are used as in situ bio-indicators of water quality due to their ability to accumulate chemicals including heavy metals<sup>37</sup>. Macrophytes take up heavy metals from the growing medium either passively, through mass flow of water in the roots or by active transport through epidermal cells of roots<sup>38</sup>. Under natural circumstances macrophytes can accumulate toxic metals to the magnitude several times greater than those in the substrate<sup>2,39</sup>. The metal accumulations by different macrophytes studied are presented in Table 5. The metal accumulation efficiencies of different macrophytes as observed are: E. crassipes > Pistia stratiotes > Polygonum coccineum > M. quadrifolia for As, Cd and Cu, whereas, Pb was accumulated maximum by Pistia stratiotes followed by Polygonum coccineum, E. crassipes and M. quadrifolia on an average. The concentration of metals recorded in different macrophytes is far above the concentration of respective metals in water. Among the different macrophytes, Pistia stratiotes and Polygonum coccineum have accumulated maximum amount of Pb followed by Cu, Cd and As. In E. crassipes and M. quadrifolia the relative metal accumulation pattern was found as Cu > Cd > Pb > As and Cu > Pb > Cd > As respectively. Similar metal accumulation patterns have been reported in Myriophyllum spicatum growing in contaminated water sources in Egypt<sup>33</sup>, E. crassipus, Ceratophyllum demersum, Typha domengensis growing naturally in the waters of river Nile<sup>28</sup>, and *E. crassipus*, Pistia stratiotes, M. quadrifolia grown in an anthropogenic lake in India<sup>18</sup>.

Translocation factor (TF) expresses the ability of a plant to transfer the metal from roots to its aerial part and is defined as the ratio of metal concentration in the shoot with that in the roots<sup>38</sup>. In the present study variable TF was observed among the different macrophytes (Table 6). It was reported that TF value greater than one (>1) was observed in metal accumulator species, while less than one (<1) was observed in metal excluder species<sup>2</sup>. Results reveal that none of the macrophytes has TF > 1, which indicates that metals are mainly stored in roots which is a basic characteristic of rhizo-filtration<sup>40</sup>. Other studies have also reported a higher concentration of metals in the roots of aquatic macrophytes<sup>2,41</sup>.

Enrichment coefficient (EC) expresses the metal accumulation efficiency of a plant with respect to its concentration in the growing medium<sup>2</sup>. For better understanding the metal storage within the plants, EC was separately calculated for root (EC<sub>R</sub>) and shoot (EC<sub>S</sub>) (Table 5). The variation in EC<sub>S</sub> and EC<sub>R</sub> of different macrophytes at different sites may be attributed to variation in physicochemical characteristics of water. Further, variation in plant genotype, active/passive transfer processes within the plant, sequestration and speciation, type and nature of roots system and preferential availability of different metals may have been responsible for heterogeneous EC<sup>42,43</sup>.

#### Conclusion

The present geochemical study of Gomti river water, river bed sediment and macrophytes suggested that the river water is polluted along the whole stretch of the river in Lucknow city. Various anthropogenic activities in and around the city contribute to the pollution. Low level of DO indicates the presence of various organic wastes in the river Cd, Cu and Pb in the water, sediments and macrophytes have been reported earlier as well, but arsenic in water and sediments has been reported for the first time. The concentration of metals in the river bed sediment was many times higher than that in river water. The river sediments not only act as depot of metals, but also act as secondary source of metals to the water and aquatic biota. Although the concentration of Cd, Cu, Pb and As is low in water, due to continuous exposure of aquatic plants to the river water, the metals are bio-accumulated in Pistia stratiotes, E. crassipes, Polygonum coccineum and *M. quadrifolia* at levels beyond the toxicity three\_old for many plants. These macrophytes provide food and shelter t other aquatic life. Hence there is increased probability of metal transfer to higher trophic levels by entering into the food chain. As the river water is being used for irrigation by local farmers, the risk of metal toxicity to man and animals cannot be ignored. In the coming years, increase in population will result in additional discharge of treated and untreated municipal and industrial effluents into the river which will further exacerbate the problem.

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