

ORIGINAL PAPER

A study of effect of storage condition on blood alcohol concentration in living subjects

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ABSTRACT

Introduction: Many studies have demonstrated that both generation and loss of alcohol in stored blood samples. Studies have concluded that both high temperature and an insufficient enzyme inhibitor concentration can result in alcohol generation, presumably as a result of bacterial fermentation. Factors most affecting the stored blood samples, to be used for ethanol determination were the duration and temperature of storage and concentration of preservative. **Objectives:** To determine whether alcohol is generated or lost in blood samples stored at different periods and with the presence or absence of preservative and refrigeration. **Methods:** 40 adult males of drunkenness were selected as subjects for estimation of blood alcohol. Then after taking written informed consent 30ml of blood is collected from the each individuals. 5 samples were preserved in sodium fluoride vacutainer and refrigerated at 4°C. Another 5 samples were preserved in plain vacutainer and kept at room temperature. Then these were subjected to estimation of blood alcohol concentration (BAC) by Gas Chromatography-Flame Ionization Detector (GC-FID) at various interval of time i.e., on 2nd, 7th, 14th, 30th and 60th day. **Conclusion:** BAC in samples without preservative and without refrigeration has fallen significantly as storage period increases at each point of time compared to BAC with preservative NaF (Sodium Fluoride) and with refrigeration at 4°C. NaF and refrigeration of samples at 4°C significantly prevents loss of BAC in stored samples.

Keywords: Blood Alcohol Concentration; Sodium Fluoride

INTRODUCTON

The accurate determination of alcohol concentration levels in human blood samples is important for valid results in research studies and often has critical medical and legal ramifications in forensic and toxicological reports.¹ Many studies have demonstrated that both generation and loss of alcohol in stored blood samples. Studies have concluded that both high temperature and an insufficient enzyme inhibitor concentration can result in alcohol generation, presumably as a result of

bacterial fermentation.² Factors most affecting the stored blood samples, to be used for ethanol determination were the duration and temperature of storage and concentration of preservative.³ Ethanol losses in samples are positively correlated with the length of storage and the original ethanol concentration in the blood.⁴ Antemortem blood samples stored at room temperature or higher will cause a decrease in BAC, not an increase.⁵

Post-mortem production of ethanol up to 70 mg% till 7th day and in few cases even up to 14th day. After 14th day there is loss of ethanol that further decreased on 28th day to become alcohol free.⁶ Majority of cases showed higher number of BAC till 20 days thereafter from 21 days to 30 days they found subsequent decrease in BAC.⁷ The possibility of investor synthesis of ethanol in samples has been raised, as well as loss due to evaporation or adsorption of the ethanol onto rubber stopper.⁸ Increase in post-mortem ethanol production is due to the presence of bacteria. More than 50 species of bacteria, yeast and fungus were capable of producing post-mortem ethanol.⁹ However in some studies freshly collected blood samples have shown that concentrations do not change in preserved samples stored in room temperature for up to two months or refrigerates samples stores up to 10 months.¹⁰

OBJECTIVE

To determine whether alcohol is generated or lost in blood samples stored at different periods and with the presence or absence of preservative and refrigeration.

METHODOLOGY

40 adult males who were brought by the police to the emergency

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department for drunkenness examination were selected as subjects for estimation of blood alcohol. Then after taking written informed consent 30ml of blood is collected from the individuals who have consumed alcohol 2 hours prior. Then the collected 30ml blood sample was equally divided into 10 parts of 3ml each, out of which 5 samples were preserved in sodium fluoride vacutainer which contained 3mg NaF per ml of blood. Samples were well mixed and refrigerated at 4°C. Another 5 parts are preserved in plain vacutainer and kept at room temperature. Then these were subjected to estimation of blood alcohol concentration (BAC) by Gas Chromatography-Flame Ionization Detector (GC-

FID) at various interval of time i.e., on 2nd, 7th, 14th, 30th and 60th day. Institutional Ethical committee clearance was obtained prior to conducting this study.

RESULTS

Tables 1 depicts BAC with preservative and with refrigeration and BAC without preservative and without refrigeration (comparison of mean differences of BAC in two groups at individual points of time) shows significant changes i.e., BAC in samples without preservative and without refrigeration had fallen significantly as storage period increases at each point of time compared to BAC with preservative and with refrigeration.

Table 1 Comparison of mean differences of BAC in two groups at individual points of time

BAC at end of	Group	Number of samples	Mean BAC	P value
2 nd Day	With Preservative and with Refrigeration	40	57.21	0.001
	Without Preservative and Without Refrigeration	40	47.09	
	Total	80		
7 th Day	With Preservative and with Refrigeration	40	51.45	0.0003
	Without Preservative and Without Refrigeration	40	42.50	
	Total	80		
14 th Day	With Preservative and with Refrigeration	40	55.84	0.0089
	Without Preservative and Without Refrigeration	40	49.20	
	Total	80		
30 th Day	With Preservative and with Refrigeration	40	49.20	0.0168
	Without Preservative and Without Refrigeration	40	44.40	
	Total	80		
60 th Day	With Preservative and with Refrigeration	40	41.92	0.0177
	Without Preservative and Without Refrigeration	40	34.68	
	Total	80		

Mann-Whitney Test the two-tailed P value is considered significant.

Table 2 depicts, when BAC samples were analysed within the groups i.e., samples with preservative and with refrigeration, we found that BAC had fallen significantly as storage time increased in this group.

Table 2 BAC (in mg %) level with Preservative and with Refrigeration at 2nd, 7th, 14th, 30th and 60th day

BAC at end of	Mean Rank
2 nd at end of	4.08
7 th Day	2.78
14 th Day	3.95
30 th Day	2.80
60 th Day	1.40

Chi Square 75.34, Degree Freedom 4

Friedman Test: The P value is < 0.0001, considered extremely significant.

Table 4 depicts, Mean BAC in samples with preservative and with refrigeration, in which Mean BAC had fallen gradually as storage period increased except on the 14th day where Mean BAC increased and then fell gradually.

Table 3 depicts, when BAC samples were analysed within the groups i.e. samples without preservative and without refrigeration, we found that BAC had fallen significantly as storage time increased in this group.

Table 3 BAC (in mg %) level without Preservative & without Refrigeration at 2nd, 7th, 14th, 30th and 60th day

BAC at end of	Mean Rank
2 nd at end of	3.50
7 th Day	2.70
14 th Day	4.00
30 th Day	3.08
60 th Day	1.73

Chi Square 47.54, Degree Freedom 4

Friedman Test: The P value is < 0.0001, considered extremely significant.

And Mean BAC in samples without preservative and without refrigeration in which Mean BAC has fallen gradually as storage period increased except on 14th day where Mean BAC had increased and then fell gradually.

Table 4 Mean BAC (in mg %) level with Preservative & with Refrigeration v/s Mean BAC (in mg %) level without Preservative & without Refrigeration

Days	Mean BAC in Sample with Preservative & with Refrigeration	SD in Sampe with Preservative & with Refrigeration	Mean BAC in Sample without Preservative & without Refrigeration	SD in Sampe without Preservative & with Refrigeration
2nd Day	57.21	11.47	47.09	11.08
7th Day	51.45	12.13	42.50	6.35
14th Day	55.84	9.26	49.20	11.03
30th Day	49.20	13.08	44.40	10.74
60th Day	41.92	10.72	34.68	14.75

DISCUSSION

BAC at 2nd, 7th, 14th, 30th and 60th day in Samples with preservative and with refrigeration v/s Samples without preservative and without refrigeration

BAC with preservative and with refrigeration and BAC without preservative and without refrigeration which shows significant changes i.e., BAC in samples without preservative and without refrigeration have fallen significantly as storage period increases at each point of time compared to BAC with preservative and with refrigeration.

The highlight of our study is use preservative NaF & refrigeration of samples at 4°C for analysing BAC in which fall of BAC is significantly less than those samples without preservative and without refrigeration.

Lewis RJ et al.¹¹ concluded that NaF and refrigeration is preferred way of storage for estimation of blood alcohol concentration of the blood samples.

In the study conducted by dubowskietal.¹² the samples which were analysed without preservative and without refrigeration showed a decrease in the BAC and when they studied the samples with preservative (NaF and biocide sodium aside) and with refrigeration there was no significant change in BAC. So also Brown et al.³ in his study concluded that factors most affecting the stored blood samples were duration, temperature of storage and concentration of preservative. Similarly Wigmore JG⁵ in his study concluded that the most accurate determination of BAC is from the blood samples which are refrigerated and Ma Dong¹³ concluded the best condition for keeping ethanol stable in blood is refrigeration with preservative and with 50% of air chamber in container. Also the results of Slavka et al.¹⁴ showed that alcohol concentrations were significantly reduced with the increase of temperature and prolongation of storage. Room temperature storage of samples is the least suitable way of keeping them, independent of the duration of storage. The temperature of storage, duration of storage, selection of preservatives and air quantity above the sample are said to be the most common causes of changes in the value of ethanol in whole blood samples. There is the synergism of these influences and it is hard to discuss the conditions separately. Wichai Wong chanapai¹⁵ concluded that the concentrations of ethanol in bloods with 1% sodium fluoride as preservative stored at 4°C were more stable than at -20°C and room temperature.

BAC at 2nd, 7th, 14th, 30th and 60th day in Samples with preservative and with refrigeration

When BAC samples analysed within groups i.e., samples with preservative and with refrigeration, we found that BAC has fallen significantly as storage time increased in this group.

AND BAC at 2nd, 7th, 14th, 30th and 60th day in Samples without preservative and without refrigeration

When BAC samples analysed within groups i.e. samples without preservative and without refrigeration, we found that BAC has fallen significantly as storage time increased in this group.

BAC in above mentioned both groups i.e. samples with preservative NaF and with refrigeration at 4°C and samples without preservative and without refrigeration has significantly fallen gradually as storage period increased.

Reason for loss of BAC in above mentioned both groups can be attributed to the chemical oxidation of the stored samples as well as due to the evaporation and adsorption.

Jones AW¹⁶ showed that ethanol losses in samples are positively correlated with the length of storage and the original ethanol concentration in the blood. Moynham et al.¹⁷ found that in blood taken from living subjects, there was no alcohol generation regardless of varying storage temperatures, times and the presence or absence of an enzyme inhibitor, but there was some alcohol depletion after longer storage times. Shan X et al.¹⁸ found that alcohol positive cases showed various changes in BAC ranging from no significant change to a 47% decrease and concluded long term storage either under refrigeration, at or above room temperature decreased BAC. Tracey Winek¹⁹ inferred that whole blood samples stored for 35 days at 26.7°C to 37.8°C lost alcohol and the percentage loss of BAC averaged between 10-19%. And important mechanism with regard to stability of alcohol in stored blood was a strongly temperature dependent alcohol oxidation reaction which was not inhibited by sodium fluoride. Avbel AJ.²⁰ showed blood samples without preservative stored under refrigeration (3°C) for 18 months to 2 years, showed decrease in ethanol content. The decrease were attributed to oxidation and (or) evaporation. Slavka Mandic-Radic¹⁴ showed that alcohol concentrations were significantly reduced with the increase of temperature and prolongation of storage. Wichai Wong chanapai¹⁵ concluded that the loss of ethanol in stored whole blood sample was due to the chemical oxidation rather than the physical loss. Anderson SG et al.² found that consistently

higher rates of alcohol depletion in the preserved samples might reflect salting-out effect and/or some reaction alcohol and sodium fluoride. Dubowski et al.¹² showed that ethanol levels in whole blood samples stored up to 1 year (refrigerated at 4°C) without preservative declined slightly (less than 5%), but this decrease was not statistically significant. Samples stored with the preservative and biocide sodium aside did not show any ethanol degradation over the 12-month storage period. Glendenning BL and Waugh TC¹⁰ concluded freshly collected blood samples have shown that concentrations do not change in preserved samples stored in room temperature for up to two months or refrigerated samples stores up to 10 months. Charies L et al.⁸ concluded in their study that Alcohol analyses of blood obtained aseptically from living humans can be delayed for as long as 14 days without a significant change in alcohol content. This hold true whether the blood sample is refrigerated or not, or whether a preservative is added to sample or not.

Mean BAC in with preservative and with refrigerated samples v/s Mean BAC in without preservative and without refrigeration samples

Mean BAC in samples with preservative and with refrigeration, in which Mean BAC has fallen gradually as storage period increased except on the 14th day where Mean BAC is increased and then fell gradually.

Mean BAC in samples without preservative and without refrigeration in which Mean BAC has fallen gradually as storage period increased except on 14th day where Mean BAC has increased and then fell gradually.

In above mentioned both groups mean increase in BAC at 14th day could be attributed to microbial fermentation due to contamination.

C B Jani et al.⁷ concluded that majority of cases showed higher number of BAC till 20 days thereafter from 21 days to 30 days they found subsequent decrease in BAC in antemortem sample. Avbel AJ²⁰ studied post-mortem human blood samples without preservative stored under refrigeration (3°C) for 18 months to 2 years, observing increase and decrease in ethanol content. The decrease were attributed to oxidation and (or) evaporation, the increases to post-mortem synthesis of ethanol by microbial fermentation of glucose. Anderson SG et al.² concluded that both high temperature and an insufficient enzyme inhibitor concentration can result in alcohol generation, presumably as a result of bacterial fermentation. Stojan Pet kovic et al.²¹ confirmed that the absences of preservative and prolonged storage at higher temperatures are not necessarily sufficient for alcohol production in antemortem blood samples. Singh and Chandra⁶ have reported that on 14th day of analysis there is post-mortem loss of ethanol that further decreased on 28day to that maximum to become alcohol free. However they also reported that maximum production of ethanol occur as 70mg % mostly within 7th day.

Usually the samples for BAC estimation which were stored without preservative at room temperature were analysed after longer duration (approximately 1 to 2 months), which lead to significant loss of BAC as the storage period increased. We recommend that samples for BAC estimation should be ideally preserved in sodium fluoride vacationer with refrigeration at 4°C.

CONCLUSION

Present study concludes BAC in samples without preservative

and without refrigeration has fallen significantly as storage period increases at each point of time compared to BAC with preservative NaF and with refrigeration at 4°C. NaF and refrigeration of samples at 4°C significantly prevents loss of BAC in stored samples.

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